with sodium acetate (-15 °C for 30 min and room temperature overnight), the Z isomer (IVa) was obtained in 69% overall yield (stereoselectivity 88%).¹⁴ Similar reaction of 5-triethylsilyl-4-decanone (Ic) with methyllithium gave IIIa in 76% vield (stereoselectivity 91%) by treatment with potassium tert-butoxide and IVa in 57% overall yield (stereoselectivity 90%) by treatment with glacial acetic acid saturated with sodium acetate.15



Further information about the stereoselectivity of this olefin synthesis was obtained from the reaction of 3-trimethylsilyl-2-octanone (Ib) with propyllithium. Treatment of the reaction mixture with potassium tert-butoxide gave IVa in 25% yield (stereoselectivity 96%) via IIb.¹⁶

When methylmagnesium bromide, in place of methyllithium, was treated with β -ketosilane Ia, IVa was obtained in 22% (stereoselectivity 77%) after acidic workup (glacial acid saturated with sodium acetate). Reaction of Ia with phenyllithium gave 4-phenyl-4-decene (52% overall yield, 39/61 E/Z) after treatment of the reaction mixture with concentrated sulfuric acid.17

These data show that β -hydroxysilanes are produced from β -ketosilanes and alkyllithium reagents in high stereoselectivity. The potassium salt of the alcohol produces a trisubstituted ethylene by syn-elimination of trimethylsilyl and hydroxy groups whereas treatment with acid in the presence of nucleophile induces anti-elimination of these two groups.1-3,6

Preparation of 4-methyl-4(E)-decene (IIIa) and the Z isomer (IVa) are representative. A stirred solution of 5-trimethylsilyl-4-decanone (Ia, 0.23 g, 1 mmol) in 5 ml of THF was treated with methyllithium (3 mmol, 3.5 ml of 0.85 M ethereal solution) at -78 °C and the reaction mixture was stirred at room temperature overnight. Addition of 1.0 g (9 mmol) of potassium tert-butoxide and refluxing of the reaction mixture for 1 h gave IIIa (74% overall yield, 91/9 E/Z): NMR (CCl₄) δ 0.6–1.04 (6 H, m), 1.04–1.5 (8 H, m), 1.55 (3 H, br s), 1.70–2.20 (4 H, m), 5.05 (1 H, br t, J = 7 Hz). When the reaction mixture from Ia and methyllithium (-78 °C for 15 min, then room temperature for 1 h) was treated with 10 ml of glacial acetic acid saturated with sodium acetate under stirring at -15 °C for 30 min and at room temperature overnight, workup gave IVa (69%, 12/88 E/Z): NMR (CCl₄) δ 0.88 (6 H, t, J = 6 Hz), 1.03–1.57 (8 H, m), 1.63 (3 H, br s), 1.70–2.17 (4 H, m), 5.00 (1 H, t, J = 7 Hz); MS m/e (rel intensity, %), 154 (M⁺, 11), 111 (13), 97 (22), 84 (16), 69 (52), 55 (100).

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- (10) Prepared from trimethylvinylsilane, butyllithium, and butyraldehyde analogously to the reported procedure (ref 2) for the preparation of β -ketosilanes. This ketosilane was also prepared from trimethylvinylsilane, butyl lithium, and butyryl chloride in 45% overall yield: bp 110-120 °C (23 mm).
- (11) Compound IIa was not isolated. Stereochemistry of IIa as well as its corresponding alcohol could not be determined but estimated from the stereochemistry of olefins prepared by syn- and anti-elimination.
- The structure of Illa and IVa are based on the NMR. The chemical shift of (12)7-methyl signal in 3-propyl-7-methyl-2(Z),6(E)-decadien-1-ol is recorded to be at δ 1.57 ppm, whereas that of the $\delta(Z)$ isomer is recorded to be at 1.66 (ref 13). The observed chemical shift of vinylic methyl of Illa appeared at δ 1.55 ppm and that of the Z isomer (IVa) at 1.63.
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- (15) Further attempts for preparation of olefins: treatment with thioglycolic acid, 61% yield, 46/54 E/Z; BF₃-OEt₂, 82, 37/63. Treatment of the reaction mixture with ammonium chloride gave β -hydroxysilane (lla', Li = H) which gave IVa: treatment with AcOH saturated with ACONa, 46% yield, 10:90 E/Z; AcOH-KF, 72, 12/88; H₂SO₄-KF, 84, 18/82; H₂SO₄, 69, 24/76. (16) Propyllithium may abstract proton from methyl ketone in lb. This side re-
- action may lower the overall yield of IVa
- Treatment of the reaction mixture with potassium tert-butoxide or glacial acetic acid saturated with sodium acetate did not give any olefin. Sulfuric (17)acid produces a benzyl-type cation which gives 4-phenyl-4-decene by the elimination of trimethylsilyl group.

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Stereoselective Synthesis of Vinylsilanes from Alkynylsilanes by Reductive Alkylation via Hydroboration and Carbodemetalation

Summary: Hydroboration of 1-trimethylsilyl-1-alkyne with dicyclohexylborane gave 1-trimethylsilylvinylborane regioand stereoselectively whose successive treatment with methyllithium, cuprous iodide, and alkyl halides afforded 1,2-dialkylvinylsilane with strict geometry.

Sir: Organosilicon compounds have attracted much attention as versatile reagents for organic synthesis,¹ for example, vinylsilanes have been shown to be useful precursor for ketones, vinyl halides, and olefins of predictable stereochemistry.^{2,3} We wish to describe here a novel stereoselective and generally applicable procedure for the reductive 1,2-dialkylvinylsilanes with fixed configuration, which is based on regioselective hydroboration of 1-alkynylsilane to vinylborane and the following stereoselective transmetalation and carbodemetalation.4

Hydroboration of 1-trimethylsilyl-1-octyne (I, R = n- C_6H_{13}) with excess dicyclohexylborane gave vinylborane II $(R = n - C_6 H_{13})$ regioselectively.^{5,6} Excess borane was quenched with 1-butene and the resulting mixture was treated with





Table I. Stereoselective Synthesis of 1,2-Dialkylvinylsilanes (IV)

^a Method B means carbodemetallation via borate and method Cu indicates carbodemetallation via vinylcopper. ^b Isolated yield after column chromatography unless otherwise stated. ^c Analysis by GLC (3×3 mm glass column packed with 20% silicon HVSG or 20% PEG 20M on Chromosorb W-AW). ^d Determined by GLC. ^e Isolated by column chromatography followed by distillation. ^f Prepared according to the modified Julia method: J. P. McCormick and D. L. Barton, J. Chem. Soc., Chem. Commun., 303 (1975). & Trimethyl phosphite was used in the place of triethyl phosphite. h Determined by NMR. ⁱ After isolation of the protected alcohol by column chromatography (silica gel or basic alumina), the alcohol was obtained by methanolysis (5 mM TsOH in MeOH, room temperature, 1 h) followed by distillation (Kugelrohr). *i* After removal of the protecting group, the alcohol was purified by column chromatography (silica gel). k 2-Methyl-2-(3-iodopropyl)-1,3-dioxolane.

methyllithium (equimolar to used borane and an additional 1 mol) and cuprous iodide affording α -silylated vinylcopper.⁷ Treatment with a wide variety of alkyl halides afforded stereoselective 1,2-dialkylvinylsilanes IV in excellent yield. The reactivity toward tosylate and homoallyl halide indicated that the above-described α -silvlated vinylcopper was more reactive than the simple vinylcopper.8 Table I shows the chemical yield and stereoselectivity of the vinylsilanes.

Preparation of 3-trimethylsilyl-3(Z)-decene is representative. To a stirred suspension of dicyclohexylborane in THF, prepared from $6 \text{ mmol} (5.6 \text{ ml of } 1.08 \text{ M solution of BH}_3 \text{ in THF}) \text{ of borane and } 0.98$ g (12 mmol) of cyclohexene in 4.0 ml of THF at 0 °C, was added 0.55 g (3.0 mmol) of 1-trimethylsilyl-1-octyne under argon. After stirring at room temperature for 5 h, the remaining dicyclohexylborane was quenched with 1-butene at 0 °C and transformed into dicyclohexylbutylborane. The reaction mixture was first treated with 9.0 mmol of methyllithium (6.6 ml of 1.36 M ethereal solution), stirred at room temperature for 20 min, and finally treated with cuprous iodide (0.57 g, 3.0 mmol) at -30 °C for 5 min. The resulting dark brown mixture was added with triethyl phosphite (0.60 g, 3.6 mmol) and hexamethylphosphoric triamide (3 ml) then with ethyl iodide (0.70 g, 4.5 mmol) at -30 °C. The reaction mixture was allowed to warm to room temperature overnight and treated with 4 ml of 3 N NaOH and 8 ml of 30% H₂O₂ at 0 °C. After stirring at room temperature for 1 h, the reaction mixture was extracted with ether, washed, dried, and chromatographed on silica gel (hexane), affording 0.56 g (88%) of IV (R $= n \cdot \tilde{C}_6 H_{13}; R' = C_2 H_5).$

Analogous to the previously reported synthesis of vinylsilane from alanates,⁴ intermediary borates, prepared from 1trimethylsilyl-1-alkyne, dicyclohexylborane, and methyllithium, could react with methyl iodide and allyl halides in excellent yield. 9,10

The above-described procedures, coupled with the facile exchange of trimethylsilyl into hydrogen,^{2b,3} provide highly stereoselective synthesis of olefins from 1-alkyne and are novel additions to the synthetic reactions with organoboron compounds.

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- NMR of the crude product showed a single olefinic proton (CCl₄, Me₄Si as internal standard, δ 5.48 ppm, t, J = 7 Hz). This observation as well as the yield of carbodemetalation product suggested that 1-trimethylsilylvinylborane II is the sole product.
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